Quiz 9 3 pts of extra credt

1. Use the tripeptide below to answer the following questions.

a) Draw an arrow pointing to each of the peptide bonds.

see above

(1 pt)

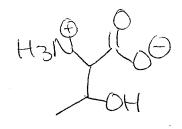
b) Use the 3-letter abbreviations to write the amino acid sequence.

(1 pt)

Thr - Phe- Lys

c) Draw the hydrolysis products of the tripeptide at physiological pH.

(4 pts)



H3N (00)

Match the isomer to its definition: CI = constitutional isomer; CF = conformer;

Han I CE

d) Star the chiral carbons in the tripeptide.

(2 pts)

see above

(1 pt)

same chemical formula with different connections between atoms

Same compound shown in different rotations around tetrahedral carbons

cis/trans isomers

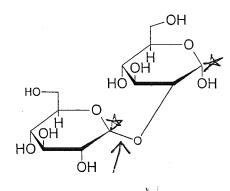
3. Classify the 3° interactions below as salt bridge (SB), London dispersion (LD), H-bonding (HB), or metal-ion complexes (MI). (2 pts)

4. Acetylcholine (Ach) is a neurotransmitter responsible for the control of skeletal muscles. After acetycholine has triggered a muscle movement, it is deactivated by the enzyme acetylcholinesterase to avoid over stimulation of the muscle. The reaction is shown below.

- a) What type of reaction does acetylcholine undergo hydration, dehydration, oxidation, reduction, synthetic dehydration, or hydrolysis? (1 pt)
- b) Nicotine interacts with acetylcholinesterase at the active site for acetylcholine. Is nicotine a competitive or non-competitive inhibitor for the reaction above? (1 pt)

Quiz 10

1. Kojibiose is a disaccharide. Use the structure of kojibiose below to answer the following questions. (5 pts)



- a) Draw an arrow pointing to the glycosidic bond.
- b) Star the anomeric carbons.
- c) This disaccharide is made from two molecules of the same monosaccharide. What is the name of this monosaccharide?

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 HMAL + |
- d) Draw the Fischer projection for the monosaccharide the forms kojibiose.



- e) Classify the glycoside bond using the α or β -(#, #) format.
- f) Is kojibiose a reduce sugar? Explain.

ges, the monosecchanide (glucose) on the right can open a become oxidized. +1

	CHEMICAL FORMULA for inorganic compounds & SKELETAL-LINE STRUCTURE for organic compounds	chemical name (IVPAC for organic compounds)
other	hes CH3	cis-3-methylcyclohexanol
	HBr	hydrobromic
	Ca (H(O3)2	calcium bicarbonate
	Pb3(PO4)2	lead(II) phosphate
	NAR	3-methylhexanal
	BąSO ₄	barium sulfate
 	HNCH3 CH3	N,2-dimethyl-3-pentanamine



Take Home Quiz - Due at the break of our next lecture session.

1. The results from testing five different unknown sugar solutions are summarized in the table below. The possible identities of the sugars are ribose, glucose, fructose, lactose, sucrose, and starch.

Use the test results and flow chart to identify the unknown sugars. Record your answers in the table below.

(5 pts)

Unknown	Benedict's	Barfoed's	Bial's	Seliwanoff's	lodine
1	blue	blue	yellow	yellow	purple
2	red ppt	red ppt	purple	yellow	yellow
3	blue	blue	yellow	yellow	yellow
4	red ppt	red ppt	green-brown	red	yellow
5	red ppt	Ыие	yellow	yellow	yellow

	Benedict's test			
	red precipitate reducing carbohydrate Barfoed's test		nydrate	
Barfoed's			lodine test	
red precipitate monosaccharide Bial's test blue or purple green-brown pentose hexose Seliwanoff's red ketohexose	no precipitate disaccharide test rellow or faint pink other hexoses	purple starch	not purple not starch	
KGIOHOXOGO	Other Hexeses	•		

Unknown	Identity
1	starch
2	ribose
3	sucrose
4	Fructose
5.	lactose

