

**COMMON IONS FOR CHEM 309**

1 +		2 +		3 +		4 +	
Ammonium	$\text{NH}_4^+$	Beryllium	$\text{Be}^{2+}$	Aluminum	$\text{Al}^{3+}$	Lead (IV) <i>(Plumbic)</i>	$\text{Pb}^{4+}$
Hydrogen	$\text{H}^+$	Magnesium	$\text{Mg}^{2+}$	Iron (III) <i>(Ferric)</i>	$\text{Fe}^{3+}$	Tin (IV) <i>(Stannic)</i>	$\text{Sn}^{4+}$
Lithium	$\text{Li}^+$	Calcium	$\text{Ca}^{2+}$				
Sodium	$\text{Na}^+$	Strontrium	$\text{Sr}^{2+}$				
Potassium	$\text{K}^+$	Barium	$\text{Ba}^{2+}$	Gold (III)	$\text{Au}^{3+}$		
Rubidium	$\text{Rb}^+$	Zinc	$\text{Zn}^{2+}$	Chromium (III)	$\text{Cr}^{3+}$		
Cesium	$\text{Cs}^+$	Copper (II) <i>(Cupric)</i>	$\text{Cu}^{2+}$				
Silver	$\text{Ag}^+$	Iron (II) <i>(Ferrous)</i>	$\text{Fe}^{2+}$				
Copper (I) <i>(Cuprous)</i>	$\text{Cu}^+$	Lead (II) <i>(Plumbous)</i>	$\text{Pb}^{2+}$				
		Mercury (I) <i>(Mercurous)</i>	$\text{Hg}^{2+}$				
		Mercury (II) <i>(Mercuric)</i>	$\text{Hg}^{2+}$				
		Tin (II) <i>(Stannous)</i>	$\text{Sn}^{2+}$				

-1		-2		-3	
Fluoride	$\text{F}^-$	Oxide	$\text{O}^{2-}$	Nitride	$\text{N}^{3-}$
Chloride	$\text{Cl}^-$	Sulfide	$\text{S}^{2-}$		
Bromide	$\text{Br}^-$			Phosphate	$\text{PO}_4^{3-}$
Iodide	$\text{I}^-$	Carbonate	$\text{CO}_3^{2-}$		
Hydroxide	$\text{OH}^-$	Hydrogen phosphate <i>(Biphosphate)</i>	$\text{HPO}_4^{2-}$		
Cyanide	$\text{CN}^-$				
Acetate	$\text{CH}_3\text{CO}_2^-$ or $\text{C}_2\text{H}_3\text{O}_2^-$	Peroxide	$\text{O}_2^{2-}$		
Nitrate	$\text{NO}_3^-$				
Nitrite	$\text{NO}_2^-$	Sulfate	$\text{SO}_4^{2-}$		
Hypochlorite	$\text{ClO}^-$	Sulfite	$\text{SO}_3^{2-}$		
Hydrogen carbonate	$\text{HCO}_3^-$ <i>(Bicarbonate)</i>				
Hydrogen sulfate	$\text{HSO}_4^-$ <i>(Bisulfate)</i>				
Di hydrogen phosphate	$\text{H}_2\text{PO}_4^-$				